

1 bbl STO - stock tank oil, GE - gas equivalent

$$GE_{STO} = n \cdot \frac{10.732 \cdot (60 + 460)}{14.7}$$

must find number of lb mol in 1 bbl STO

$$n_{STO} = 350.54 \cdot \frac{Y_{STO}}{M_{STO}} \quad (\text{lb mol})$$

Inserted into GE_{STO}

$$GE_{STO} = V_{sc} = \frac{Y_{STO}}{M_{STO}} \cdot 350.54 \cdot \frac{10.732 \cdot 520}{14.7}$$

For 1 bbl
STO

$$GE_{STO} \approx 133000 \frac{Y_{STO}}{M_{STO}} \frac{\text{SCF}}{\text{SBL}}$$

Field units

$$1 \text{ Sm}^3 \text{ oil} \quad GE_{STO} \approx \frac{133000}{5.615} \frac{Y_{STO}}{M_{STO}} \approx 23696 \frac{Y_{STO}}{M_{STO}} \frac{\text{Sm}^3}{\text{Sm}^3}$$

$$\text{Total } GE_{STO} (\text{SCF}) = GE_{STO} \left(\frac{\text{SCF}}{\text{1 BBL}} \right) \cdot V_{STO} (\text{SBL})$$

If only γ_{STO} is given :

(2)

Craigo's formula to find M_{STO} :

$$M_{STO} = \frac{6084}{\gamma_{API} - 5.9}$$

$$\gamma_{API} = \frac{141.5}{\gamma_o} - 131.5$$

Gas equivalent volume of produced water :

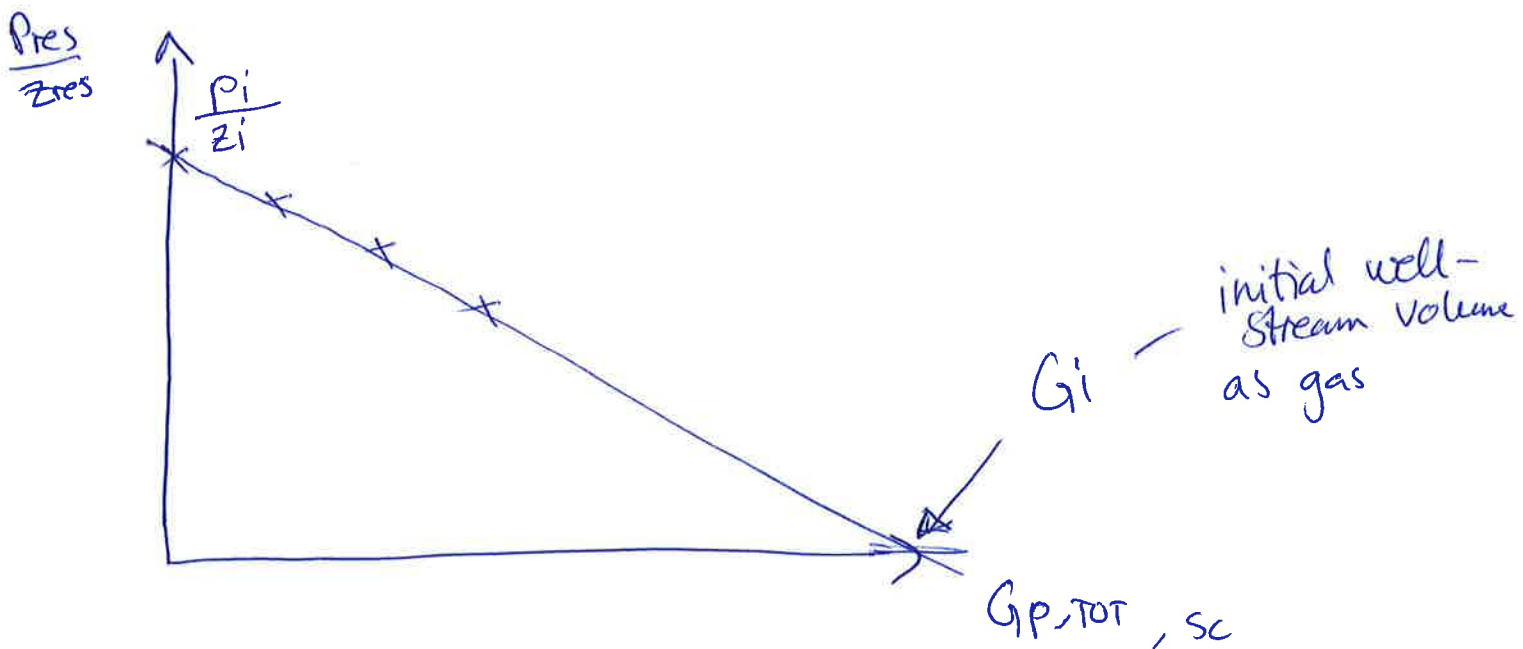
$$\gamma_w = 1 \quad M_w = 18$$

$$1 \text{ bbl} \quad G_{Ew} \approx 133000 \cdot \frac{1}{18} = 7389 \frac{\text{SCF}}{\text{SBL}}$$

Wellstream \Rightarrow gas produced :

$$Q_{p, \text{TOT}} = V_{g, \text{sc}} + V_{STO} \cdot G_{ESTO} + (V_w \cdot G_{Ew})_{\text{sc}} + (V_{\text{vap}} G_{\text{vap}})_{\text{sc}}$$

produced gas
produced oil in gas equivalents
prod. fresh water
(V_{vap} G_{vap})_{sc}



IGIP and IOIP
 ↳ initial gas in place ↳ initial oil in place

To determine IGIP and IOIP

⇒ we need to know the mol fractions of the reservoir fluid

V (gas) - mol fraction gas
 L (liquid) - mol fraction liquid.

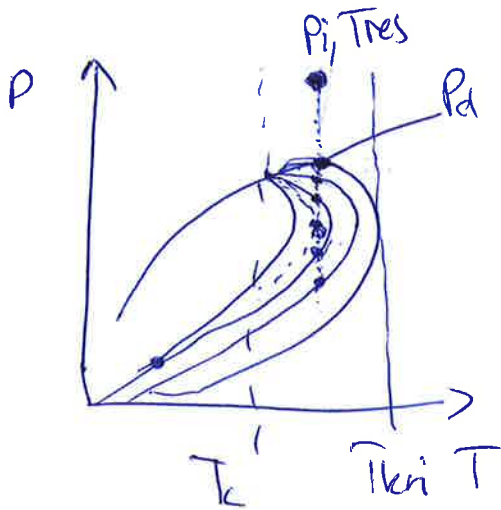
V and L can be determined in a flash process in the lab.

$$G_i = IGIP + IOIP - G_{E_{sto}}$$

$$GOR_{initially} = \frac{IGIP}{IOIP} = \frac{SCF}{SBL}$$

GAS CONDENSATES

(4)



$T_c < T_{res} < T_{kvi}$
 & inside two-phase envelope

Retrograde condensations

- liquid is condensing at lower P, liquid volume is increasing to a maximum, then evaporates again.

⊗ Calculation of oil and gas production:

Material balance ~~equation~~ calculations are depending on P_{res} :
 higher than P_d
 lower than P_d

1. $P_{res} > P_d$: Mat. bal. as for wet gas
 Convert produced STD (and water) into gas equivalents.

2. $P_{res} < P_d$: PVT-analysis to be performed in the lab.
 Production of oil and gas based on CVD-analysis data.
 "constant volume depletion"
 "konstant volumerastnings-analyse".

$P_{res} < P_d$: large amounts of liquid condensing in the reservoir, This is usually lost. \Rightarrow Productivity is reduced. (5)

Gas production is impaired by near-well blockage of condensed oil.

CVD - analysis

PVT-cell : oil and gas from separator are recombined to represent the reservoir fluid, \rightarrow added to the PVT-cell.

P_d - dew point pressure

T_{res} - reservoir temperature

Note volume of cell at these conditions.

Cell volume is increased by ΔV_1 , pressure drops to P_1 . After new equilibrium between oil and gas has been established, a volume ΔV_1 of gas is produced. Pressure still at P_1 .

Retrograde liquid volume

Compositions of produced gas and oil } determined

Z-factor for gas

Procedure is repeated approx. 10 times until abandonment pressure has been reached.

Simulating pressure depletion of a reservoir.

* Calculate produced well stream as gas at sc, $\Delta G_{p,sc}$:

The ratio between produced gas and reservoir gas volume is equal to the ratio produced gas from the PVT-cell and the volume of the PVT-cell



- Valid when P and T conditions are the same in reservoir and cell.
Also at standard conditions, sc.

For \Rightarrow pressure interval i :

$$\frac{(\Delta G_p)_i}{V_{HCPV}} = \frac{(\Delta V_g)_i}{V_{cell}} \Rightarrow (\Delta G_p)_i = V_{HCPV} \cdot \frac{(\Delta V_g)_i}{V_{cell}}$$

$(\Delta G_p)_i$ = produced well stream fluid as gas in pressure interval i

$(\Delta V_g)_i$ = produced gas volume from the cell in pressure interval i

V_{cell} = volume of gas initially in PVT-cell

V_{HCPV} = hydrocarbon pure volume
 \Rightarrow All volume are given at sc - standard conditions

$V_{HCPV,sc}$
 $(\Delta V_g)_{i,sc}$
 $V_{cell,sc}$

} can be determined by finding the number of moles in the actual volume, by $PV = ZnRT \Rightarrow n = \frac{PV}{ZRT}$, then multiplying with V_m - molar volume
 volume of 1 mol gas at standard conditions.

$$V_{HCPV,sc} = \frac{P_d \cdot V_{HCPV,res}}{Z_d \cdot R \cdot T_{res}} \cdot V_m \quad (n \cdot V_m)$$

$$(\Delta V_g)_{i,sc} = \frac{P_i \cdot (\Delta V_g)_{i,res}}{Z_i \cdot R \cdot T_{res}} \cdot V_m$$

d = dew point
 res = reservoir
 i = pressure interval

$$V_{cell,sc} = \frac{P_d \cdot V_{cell,res}}{Z_d \cdot R \cdot T_{res}} \cdot V_m$$

Use these three expressions to determine produced well stream as gas, $(\Delta G_p)_{i,sc}$, in the pressure interval i .

* How much gas and oil is produced?

GOR for each pressure step.

$$\text{GOR} = \frac{(\Delta V_g)_{i,sc}}{(\Delta V_o)_{i,sc}}$$

volume separator gas
volume STO

* Produced separator gas volume at sc
 $(\Delta V_g)_i$

For pressure interval i :

Volume of separator gas is found by subtracting the part of the well stream that becomes oil, from the ^{total} well stream as gas. Heavier components, a fraction of these, will be part of the oil phase.

$$(\Delta V_g)_{i,sc} = (\Delta G_p)_{i,sc} - (\Delta G_L)_{i,sc}$$

Well stream volume that becomes liquid.
(gas volume, SCF)

How to determine the part of the well stream that becomes liquid.

$$(\Delta G_L)_{i,sc} = (\Delta n_L)_i (\Delta G_p)_{i,sc}$$

$(\Delta n_L)_i$ = liquid mol fraction, the fraction that becomes liquid.

$(\Delta n_L)_i$ is found by multiplying the mol fractions (Z_i) for each liquid component (C_4, C_5, C_6, C_7+) with liquid recovery factors for the well stream (a, b, c, d)

\hookrightarrow fraction of the component in liquid phase.

$$(\Delta n_L)_i = a \cdot (Z_{C_4})_i + b(Z_{C_5})_i + c(Z_{C_6})_i + d(Z_{C_7+})_i$$

⊗ Produced liquid volume, $(\Delta V_o)_i$, sc

~~Calculate~~ Calculate the liquid volume, V_L , for every component, then add them together.